E

Reg No.:\_\_\_\_\_ Name:\_\_\_\_

## APJ ABDUL KALAM TECHNOLOGICAL UNIVERSITY

B.Tech Degree S2 (R) Examination May 2025 (2024 Scheme)

**Course Code: PCMET205** 

Course Name: MATERIAL SCIENCE AND ENGINEERING

| Max. | Marks | s: 60 Duration: 21  | nours30 | minutes |
|------|-------|---|---------|---------|
|      |       |   |         |         |
|      |       | PART A (Answer all questions. Each question carries 3 marks)  | СО      | Marks   |
| 1    |       | Compare the covalent, ionic and metallic bonds. Also give suitable examples.  | CO1     | (3)     |
| 2    |       | Calculate the crystal lattice parameter of a palladium FCC crystal structure, having a density of 12.0 g/cm <sup>3</sup> , and an atomic weight of 106.4 g/mol.  Avogadro number is 6.022 x10 <sup>23</sup> atoms/mol | CO1     | (3)     |
| 3    |       | What are point defects? List the different point defects in crystals.   | CO2     | (3)     |
| 4    |       | List the differences between SEM and TEM.   | CO2     | (3)     |
| 5    |       | Explain ductile-to-brittle transition temperature.  | CO3     | (3)     |
| 6    |       | How impact toughness of a material is found out? List and brief the different test methods.   | CO3     | (3)     |
| 7    |       | What is an isomorphous system? Give an example.   | CO4     | (3)     |
| 8    |       | Define tempering and List the various tempering processes.  | CO4     | (3)     |
|      |       | PART B (Answer any one full question from each module, each question carries 9 ma   | rks)    |         |
|      |       | Module -1   |         |         |
| 9    | a)    | Find the number of atoms, coordination number and atomic packing factor for HCP crystals.   | CO1     | (5)     |
|      | b)    | Classify engineering materials. List and explain their important mechanical properties.   | CO1     | (4)     |
| 10   | a)    | Draw the directions and planes corresponding to the following Miller indices.   | CO1     | (9)     |
|      |       | (a) [101], (b) [121], (c) (111), (d) (102) and (e) (220).  Write down the steps followed.   |         |         |

Module -2

## 03PCMET205052505

| 11 | a) | What are dislocations? Discuss the types of dislocations in detail with neat   | CO2 | (5) |
|----|----|--|-----|-----|
|    |    | sketches.  |     |     |
|    | b) | What is a Frank-Read Source.? With neat sketches explain the functioning       | CO2 | (4) |
|    |    | of such a source   |     |     |
| 12 | a) | State and derive the expressions for Fick's laws of diffusion for steady state | CO2 | (5) |
|    |    | and non-steady state diffusion .what are the Factors affecting diffusions      |     |     |
|    | b) | Write notes on wear and super alloys   | CO2 | (4) |
|    |    | Module -3  |     |     |
| 13 | a) | Define hardness? How hardness testing is done? Explain with a neat sketch.     | CO3 | (5) |
|    | b) | Draw engineering stress-strain diagram and explain the important points on     | CO3 | (4) |
|    |    | the curve.   |     |     |
| 14 | a) | Define creep. Explain the typical creep curve of a ferrous material. Also list | CO3 | (5) |
|    |    | out the factors affecting creep.   |     |     |
|    | b) | What is steel? Classify.   | CO3 | (4) |
|    |    | Module -4  |     |     |
| 15 | a) | Draw and explain a suitable eutectic system with complete solubility of        | CO4 | (5) |
|    |    | components in the liquid phase and partially soluble in solid phases.          |     |     |
|    | b) | Draw and explain the TTT diagram of eutectoid steel.                           | CO4 | (4) |
| 16 | a) | What is hardenability? Explain the Jominy End quench test.                     | CO4 | (5) |
|    | b) | Explain any two surface hardening methods with neat sketches.                  | CO4 | (4) |
|    |    |  |     |     |