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Reg No.: _____

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APJ ABDUL KALAM TECHNOLOGICAL UNIVERSITY

Third Semester B.Tech Degree Regular and Supplementary Examination December 2022 (2019 Scheme)



Course Code: MET205

Course Name: METALLURGY AND MATERIAL SCIENCE

Max. Marks: 100

Duration: 3 Hours

PART A

Answer all questions. Each question carries 3 marks

Marks

- 1 Describe the importance of long range and short range order of atomic arrangement on mechanical strength in solids. (3)
- 2 Write three differences between slip & twinning. (3)
- 3 What is Frank-Read source? (3)
- 4 What are the distinct features of Scanning Electronic Microscope (SEM)? (3)
- 5 What do you mean by critical cooling rate in CCT diagram? (3)
- 6 How phase amounts can be determined using phase diagram? (3)
- 7 Describe precipitation hardening mechanism in materials. (3)
- 8 List the composition, properties, and use of Gun metal. (3)
- 9 What are composites? Give one example and its application. (3)
- 10 Differentiate Creep & Fatigue. (3)

PART B

Answer any one full question from each module. Each question carries 14 marks

Module 1

- 11 a) Prove that the resolved shear stress on the slip plane in the slip direction is zero, when the slip plane is normal to the stress axis (7)
- b) Explain atomic packing factor and its importance. (4)
- c) Compare primary bond with secondary bond. (3)
- 12 a) Explain the phenomenon recovery, recrystallization and grain growth. (9)
- b) State and explain Schmid's law. (5)

Module 2

- 13 a) How X-Ray diffraction technique is utilized for the determination of crystal structures? (4)

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- b) A hardening process is performed at 1000°C on an iron-carbon alloy. The initial concentration of carbon in the alloy is 0.20 wt% C. The surface concentration is to be maintained at 1.30 wt% C. Determine the carburizing time needed to achieve a carbon concentration of 0.45 wt% at a position 2 mm. Assume the error function value 'z' as 0.85. The temperature-independent preexponential of diffusion is $2.3 \times 10^{-5} \text{ m}^2/\text{sec}$ and the activation energy is 148 kJ/mol. (10)
- 14 a) Write a short note on different types of Line defects occur in crystals. (7)
- b) Explain unsteady state diffusion. Derive the equation for Fick's second law. (7)

Module 3

- 15 a) Write the procedure for constructing TTT diagram. Mark all regions in a TTT diagram for Fe-C alloy system (10)
- b) Differentiate Normalising & Annealing. (4)
- 16 a) Why surface hardening processes are done on steels? Explain any two surface hardening methods. (10)
- b) What is invariant reaction? Compare eutectoid and peritectoid reaction. (4)

Module 4

- 17 a) How various alloying elements influence the property of steel? (7)
- b) Give the composition and applications of i) high speed steel ii) white iron (4)
- c) How cold working increases the strength of a material? (3)
- 18 a) Explain the following strengthening mechanism in metals.
- i) Strengthening by grain size reduction (5)
- ii) Solid solution strengthening (5)
- b) What are non-ferrous alloys? Give two examples and their applications. (4)

Module 5

- 19 a) With the aid of S-N curve explain the mechanism of Creep. (10)
- b) What are super alloys? List its advantages and applications (4)
- 20 a) What is fatigue failure? Explain various factors that affect the fatigue life of a material. (10)
- b) What is maraging steel? Write any two engineering applications (4)